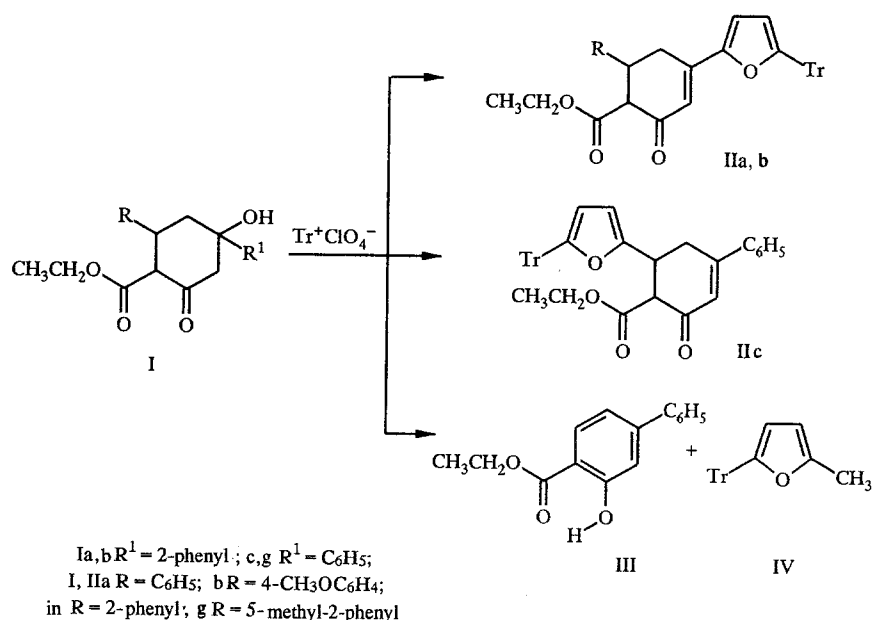


REACTION OF CYCLIC  $\beta$ -KETOALCOHOLS WITH TRITYL PERCHLORATE

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It has previously been reported [1] that 5-furyl- $\beta$ -cyclohexanones Ic, d readily undergo aromatization on acid catalysis with splitting off of furan (or sylvan) and formation of salicyclic acid derivatives. We tried to aromatize compounds Ia-d while retaining all the substituents in the ring, using trityl perchlorate as an effective hydride ions acceptor [2, 3]. Our experiments showed that the reaction of compounds Ia-d with trityl perchlorate ( $\text{TrClO}_4$ ) proceeds without splitting off of hydride ion from the nonaromatic carbocyclic ring:



In the case of compounds Ia-c containing a 5-unsubstituted furan ring, the trityl group substitutes hydrogen at the 5-position of the furan ring, while the thus liberated perchloric acid catalyzes the dehydration with the formation of compounds IIa-c. In the presence of a substituent in the 5-position of the furan ring (compound IId), trityl perchlorate acts as a 5-methylfuran ring "acceptor," which finally leads to the previously described [1] ethyl ester of p-phenylsalicyclic acid III and triphenyl(5-methyl-2-furyl)methane IV.

**5-Phenyl-3-(5-triphenylmethyl-2-furyl)-6-ethoxycarbonyl-2-cyclohexen-1-one (IIa,  $\text{C}_{38}\text{H}_{32}\text{O}_4$ ),** mp 218-220°C. IR spectrum (Nujol),  $\text{cm}^{-1}$ : 1600, 1640, 1720. PMR spectrum (60 MHz, acetone- $\text{D}_6$ ), ppm: 0.93 (3-H, t,  $\text{CH}_3\text{CH}_2\text{CO}$ ,  $J = 7.2$  Hz), 2.80 (2H, m, 4- $\text{H}_a$  and 4- $\text{H}_b$ ), 3.50 (2H, q,  $\text{CH}_3\text{CH}_2\text{CO}$ ), 3.62 (2H, m, 5-H and 6-H), 6.12 (1H, d, 4- $\text{H}_{\text{furan}}$ ,  $J = 3.6$  Hz), 6.27 (1H, d, 2-H),  $J = 2.0$  Hz), 6.73 (1H, d, 3- $\text{H}_{\text{furan}}$ ), 7.17 (15H, m,  $\text{C}(\text{C}_6\text{H}_5)_3$ ), 7.25 (5H, s,  $\text{C}_6\text{H}_5$ ).

**5-(4-Methoxyphenyl)-3-(5-triphenylmethyl-2-furyl)-6-ethoxycarbonyl-2-cyclohexen-1-one (IIb,  $\text{C}_{39}\text{H}_{34}\text{O}_5$ ),** mp 214-216°C. IR spectrum (Nujol),  $\text{cm}^{-1}$ : 1600, 1650, 1720. PMR spectrum (60 MHz, acetone- $\text{D}_6$ ), ppm: 1.1 (3-H), t,  $\text{CH}_3\text{CH}_2\text{CO}$ ,  $J = 7.1$  Hz), 2.75 (1H, m, 4- $\text{H}_a$ ), 2.80 (1H, m, 4- $\text{H}_b$ ), 3.50 (2H, m, 5-H and 6-H), 3.52, (2H, q,  $\text{CH}_3\text{CH}_2\text{CO}$ ), 3.65 (3H, s,

CH<sub>3</sub>O), 6.07 (1H, d, 4-H<sub>furan</sub>, J = 3.6 Hz), 6.28 (1H, d, 2-H, J = 2.0 Hz), 6.58 (1H, d, 3-H<sub>furan</sub>), 6.70 (2H, d, 2-H and 6-H of the benzene ring), 7.10 (2H, d, 3-H and 5-H of the benzene ring), 7.17 (15H, m, C(C<sub>6</sub>H<sub>5</sub>)<sub>3</sub>), 7.27 (5H, s, C<sub>6</sub>H<sub>5</sub>).

**3-Phenyl-5-(5-triphenylmethyl-2-furyl)-6-ethoxycarbonyl-2-cyclohexen-1-one (IIb, C<sub>38</sub>H<sub>32</sub>O<sub>4</sub>)**, mp 153-155°C. IR spectrum (Nujol), cm<sup>-1</sup>: 1600, 1640, 1720. PMR spectrum (60 MHz, acetone-D<sub>6</sub>), ppm: 1.10 (3-H, t, CH<sub>3</sub>CH<sub>2</sub>CO, J = 7.2 Hz), 2.95 (2H, m, 4-H<sub>a</sub> and 4-H<sub>e</sub>), 3.63 (2H, m, 5-H and 6-H), 4.02 (2H, q, CH<sub>3</sub>CH<sub>2</sub>CO), 5.83 (1H, d, 4-H<sub>furan</sub>, J = 3.6 Hz), 6.03 (1H, d, 3-H<sub>furan</sub>), 6.35 (1H, d, 2-H, J = 2.0 Hz), 7.12 (15H, m, C(C<sub>6</sub>H<sub>5</sub>)<sub>3</sub>), 7.35 (5H, s, C<sub>6</sub>H<sub>5</sub>).

## REFERENCES

1. E. B. Usova, G. D. Krapivin, and V. G. Kul'nevich, *Khim. Geterotsikl. Soedin.*, No. 9, 1289 (1992).
2. G. N. Dorofeenko, E. I. Sadekova, and E. V. Kuznetsov, *Preparative Chemistry of Pirylium Salts* [in Russian], Izd-vo Rostov. Univ., Rostov-on-Don (1972).
3. T. V. Stupnikova, L. A. Rybenko, and Z. M. Skorobogatova, *Khim. Geterotsikl. Soedin.*, No. 3, 416 (1978).